Articles

Topographical Amino Acid Substitution in Position 10 of Glucagon Leads to Antagonists/Partial Agonists with Greater Binding Differences

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The role of position 10 in the β -turn region of glucagon was investigated by substituting chiral constrained amino acids and other modifications in the N-terminal region. A series of glucagon analogues have been designed and synthesized by incorporating β -methylphenylalanine isomers (2S,3S, 2S,3R, 2R,3R,and 2R,3S) at position 10 in order to explore the structural and topographical requirements of the glucagon receptor, and, in addition, utilizing previous studies which indicated that antagonism could be enhanced by modifications (des-His¹, Glu⁹) and a bulky group at position 5. The structures of the new analogues are as follows: [des-His1,-Tyr⁵,Glu⁹]glucagon-NH₂ (III), [des-His¹,Tyr⁵,Glu⁹,Phe¹⁰]glucagon-NH₂ (III), [des-His¹,Tyr⁵,Glu⁹,-Ala¹⁰]glucagon-NH₂ (**IV**), [des-His¹,Tyr⁵,Glu⁹,(2S,3R)-β-MePhe¹⁰]glucagon-NH₂ (**V**), [des-His¹,-Tyr⁵,Glu⁹,(2S,3S)-β-MePhe¹⁰]glucagon-NH₂ (VI), [des-His¹,Tyr⁵,Glu⁹,D-Tyr¹⁰]glucagon-NH₂ (VII), [des-His¹,Tyr⁵,Glu⁹,D-Phe¹⁰]glucagon-NH₂ (VIII), [des-His¹,Tyr⁵,Glu⁹,D-Åla¹⁰]glucagon-NH₂ (IX), [des-His¹,Tyr⁵,Glu⁹,(2R,3R)- β -MePhe¹⁰]glucagon-NH₂ (**X**), and [des-His¹,Tyr⁵,Glu⁹,(2R,3S)- β -MePhe¹⁰|glucagon-NH₂ (XI). These analogues led to dramatically different changes in *in vitro* binding affinities for glucagon receptors. Their receptor binding potencies IC50 values (nM) are 2.3 (II), 4.1 (III), 395.0 (IV), 10.0 (V), 170.0 (VI), 74.0 (VII), 34.5 (VIII), 510.0 (IX), 120.0 (X), and 180.0 (XI). Analogues II, III, V, VI, and XI were found to be weak partial agonists/ partial antagonists with maximum stimulation between 5%-9%, while the other compounds (IV and VII-X) were antagonists unable to activate the adenylate cyclase system even at concentrations as high as 10^{-5} M. In competition experiments, all of the analogues caused a right shift of the glucagon-stimulated adenylate cyclase dose—response curve. The p A_2 values were 6.60 (II), 6.85 (III), 6.20 (IV), 6.20 (V), 6.10 (VI), 6.50 (VII), 6.20 (VIII), 5.85 (IX), 6.20 (X), and 6.00 (XI). Putative topographical requirements of the glucagon receptor for the aromatic side chain conformation in position 10 of glucagon antagonists are discussed.

Introduction

Glucagon is a 29-amino acid peptide hormone secreted by the $\alpha\text{-cell}$ in the pancreatic islets and plays a crucial role in glucose homeostasis, stimulating gluconeogenesis and glycogenolysis in the liver, leading to the production of glucose for release into the bloodstream. 1,2 Glucagon also causes lipolysis in the liver and fat cells. 3 The hormone binds to specific receptors in the liver plasma membrane followed by the activation of adenylate cyclase to increase the levels of the second messenger cAMP.

The role of glucagon in the pathogenesis of diabetes mellitus remains controversial. The bihormonal hypothesis^{4–6} postulates that in type II diabetes the overproduction of glucose is due to excess circulating glucagon, while the underutilization of glucose is realized because of insulin abnormalities. Several researchers have tried to develop and design glucagon inhibitors (antagonists) that will provide direct evidence to test this hypothesis.⁷ These glucagon receptor antagonists should bind to the plasma membrane receptor without signal transduction leading to activation of adenylate cyclase system or other second messengers.

Recently we have reported the synthesis and biological activity of a new glucagon analogue that was

designed as a potent receptor antagonist with modifications in the N-terminal region. The structure of the new analogue is [des-His¹,des-Phe⁶,Glu⁹]glucagon-NH₂, and its binding potency IC₅₀ value was 48 nM (glucagon's $IC_{50} = 1.5$ nM). This analogue was found to be the first "pure" glucagon antagonist⁸ in a highly sensitive assay for glucagon-stimulated cAMP accumulation activity and showed a p A_2 value of 8.20. The glucagon receptor antagonist was tested in vivo (manuscript submitted), where it lowered the blood glucose levels to 60% in streptozotocin-induced diabetic rats. Two other glucagon antagonists [1- N^{α} -(trinitrophenyl)histidine,12-homoarginine]glucagon (THG)⁹ and [des-His¹,Glu⁹]glucagon-NH₂,¹⁰ also were found to lower blood glucose in streptozotocin-induced diabetic rats.¹¹ These glucagon receptor antagonists reduced the blood glucose levels \sim 55-60% in diabetic animals without the addition of exogenous insulin.

Recent isolation of the glucagon receptor by Jelinek *et al.*¹² has demonstrated that it belongs to a family of G-protein-coupled hormone receptors and is related to those of calcitonin (CT), parathyroid hormone (PTH), secretin, and vasoactive intestinal peptide (VIP). The cloned receptor-bound glucagon caused an increase in the intracellular concentration of cAMP and also transduced a signal that led to an increase in calcium concentration. It has been shown that glucagon recep-

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Table 1. Structure of Glucagon and Its Analogues

peptide	structure
I, glucagon	His-Ser-Gln-Gly-Thr ⁵ -Phe-Thr-Ser-Asp-Tyr ¹⁰ -Ser-Lys-Tyr-Leu-Asp ¹⁵ -Ser- Arg-Arg-Ala-Gln ²⁰ -Asp-Phe-Val-Gln-Trp ²⁵ -Leu-Met-Asn-Thr-COOH
II	[des-His ¹ ,Tyr ⁵ ,Glu ⁹]glucagon-NH ₂
III	[des-His ¹ ,Tyr ⁵ ,Glu ⁹ ,Phe ¹⁰]glucagon-NH ₂
IV	des-His ¹ ,Tyr ⁵ ,Glu ⁹ ,Ala ¹⁰]glucagon-NH ₂
V	$[{ m des-His^1, Tyr^5, Glu^9, (2.S, 3 ilde{R})}$ - eta -MePhe $^{10}]$ glucagon-NH $_2$
VI	[des-His 1 ,Tyr 5 ,Glu 9 ,(2 S,3.S)- 2 -MePhe 10]glucagon-NH $_2$
VII	des-His¹,Tyr⁵,Glu ⁹ ,D-Tyr¹0]glucagon-NH ₂
VIII	des-His ¹ ,Tyr ⁵ ,Glu ⁹ ,D-Phe ¹⁰]glucagon-NH ₂
IX	des-His¹,Tyr⁵,Glu ⁹ ,D-Ala¹0]glucagon-NH₂
X	$[{ m des-His^1, Tyr^5, Glu^9, (2\it R, 3\it R)^-eta-MePhe^{10}}]$ glucagon-N ${ m H_2}$
XI	[des-His 1 ,Tyr 5 ,Glu 9 ,(2 R ,3 S)- β -MePhe 10]glucagon-NH $_2$

tor activation involves not only the adenylate cyclase system but also inositol phosphate. Wakelam $et\ al.^{13}$ demonstrated that glucagon at low concentration causes the production of inositol phosphates and the breakdown of phospholipids.

The major goal in this research has been to design and synthesize potent glucagon analogues that will enable us to separate the transduction message of the hormonal signal from the structural components that are necessary for binding the hormone to its receptor. From structure—activity relationship studies, it has been demonstrated that the N-terminal region of glucagon, in particular residues 1—6, 9, and 10, is important for the transduction of glucagon's physiological message, whereas the C-terminal region is important for binding to the receptor. 14.15

In this paper we describe the total synthesis of 10 new glucagon analogues by solid-phase methodology. This investigation examined modifications in the N-terminal region of glucagon, in particular residues 1, 5, 9, and 10 (histidine, threonine, aspartic acid, and tyrosine, respectively), and in addition β -MePhe isomers were introduced in the Tyr¹⁰ position of glucagon to study the effects of constraint or bias conformational preferences of the side chain moiety¹⁶ on biological activities. Since methods of asymmetric synthesis of β -methyl aromatic amino acids have been developed, 16-18 serial substitutions of biologically important aromatic amino acids with four β -methylated stereoisomers became an important tool for investigating topographical requirements of peptide receptors for the side chain conformations. 19 The β -methyl derivatives of Phe, Tyr, and Trp have been incorporated into various biologically active peptides^{20–23} to better understand structure–activity relationship. The importance of residue 1 to transduction was noted by Lin et al.24 who indicated that des-His1 was a weak partial agonist. It has been demonstrated that the position 9 aspartic acid is crucial for transduction of the message and operates in conjunction with His1 for activation of glucagon's action.25 The potent antagonist [des-His1,Glu9]glucagon-NH2 with a binding affinity of 40% clearly indicates the roles of positions 1 and 9,26,27 whereas Krystenansky et al.28 have reported the importance of the phenolic hydroxyl group of the position 10 tyrosine. Finally, the replacement of threonine by tyrosine at position 5 has been studied in this laboratory and was found to produce glucagon antagonists.^{29,30} In this study we have evaluated the possible synergy of these positions for enhancing the antagonist activity of glucagon analogues.

Results

Chemical Synthesis and Characterization. The 10 new glucagon analogues II-XI (Table 1) were synthesized by solid-phase methods using either the N^{α} -Fmoc or the N^{α} -Boc strategy. For the N^{α} -Fmoc strategy, compounds II-IV, and VII-XI were prepared using 4-[(2',4'-dimethoxyphenyl)-Fmoc-(aminomethyl)]phenoxy resin so as to obtain the carboxamide terminal peptide. Most coupling reactions employed 0.5 M HOBt/ 0.45 M HBTU in NMP. After drying the resin, the analogues were cleaved from the resin by using a cleavage mixture of 90% TFA, 5% anisole, 2.5% methyl sulfide, and 2.5% 1,2-ethanedithiol. In the N^{α} -Boc strategy, analogues V and VI were synthesized using a *p*-methylbenzhdrylamine resin substituted at the level of 0.51 mmol of NH₂/g. Briefly, most coupling reactions employed 1,3-diisopropylcarbodiimide (DIC) and Nhydroxybenzotriazole (HOBt) and (benzotriazolyloxy)tris(dimethylamino)phosphonium hexafluorophosphate (BOP) in NMP in the presence of DIEA as coupling reagents. The reactions were monitored by the Kaiser test.31 A high/low-HF technique was used to cleave the peptide from the resin. The crude peptides were isolated and purified by reversed-phase high-pressure liquid chromatography (RP-HPLC). The purity of the peptides was characterized by electrospray mass spectroscopy, thin-layer chromatography (TLC) in three different solvent systems, and amino acid analysis (see the Experimental Section).

Biological Studies. The agonist and antagonist properties of the peptides were investigated in the adenylate cyclase assay, and the results obtained are shown in Table 2. The binding potency for these compounds was obtained using liver plasma membranes in which the displacement of ¹²⁵I-labeled glucagon was measured, and the results are shown in Table 2.

In compound **II**, where Tyr^{10} was retained, binding nearly equipotent ($IC_{50} = 2.3$ nM) with that of glucagon was obtained as a result of introducing a bulkier side chain in position 5, where Tyr was substituted for Thr in conjunction with the des- His^1 and Glu^9 modifications. However, this compound turned out to be a partial agonist with a maximum stimulation of only 5%. The glucagon analogue **III** in which Phe^{10} replaced Tyr^{10} also was found to be a partial agonist with a maximum stimulation of 6% and a pA_2 value of 6.85. The binding potency IC_{50} value for this analogue was 4.1 nM (Table 2). Compounds **IV** and **IX** were found to be antagonists with less than 1% binding potency to the glucagon receptor. The results were consistent with our previous findings²⁸ and strongly support the hypothesis that

Table 2. Biological Activities of Glucagon Analogues

	receptor	binding	adenylate cyclase		
peptide ^a	IC ₅₀ (nM)	rel binding potency (%) ^b	EC ₅₀ (nM)	max stimulation (%)	pA_2
I	1.5 ± 0.20	100	8 ± 1.25	100	
II	2.3 ± 0.31	65	_	5	6.60
III	4.1 ± 1.15	37	_	6	6.85
IV	$395.\pm48.0$	0.4	$\mathbf{i}\mathbf{a}^c$	0	6.20
V	$10.\pm2.5$	15.0	_	7	6.20
VI	$170. \pm 26.$	1.0	_	9	6.10
VII	$74. \pm 20.$	2.0	ia	0	6.50
VIII	34.5 ± 7.0	4.3	ia	0	6.20
IX	$510. \pm 81.$	0.3	ia	0	5.85
X	$120. \pm 19.$	1.3	ia	0	6.20
XI	$180.\pm23.$	0.8	_	5.4	6.15

 a See Table 1 for structures. b Relative binding potency = [(receptor binding IC₅₀ for glucagon)/(receptor binding IC₅₀ for glucagon analogue)] \times 100. c ia, inactive at 10⁻⁵ M.

Tyr¹⁰ is essential for potent binding in glucagon, since the removal of the phenolic ring reduces the binding over 2 orders of magnitude (Table 2). In compounds V and **VI**, we have substituted the (2S,3S)- and (2S,3R)- β -MePhe isomers at position 10. The effect of substitution of the β -chiral center in this residue led to dramatically different binding potencies to the glucagon receptor. The binding potencies for V and VI were found to be 10 and 172 nM, respectively. The 17-fold difference in binding between these two analogues undoubtedly is due to the topographical differences caused by incorporating these asymmetric amino acids into position 10 and demonstrates the critical importance of the 10 position in glucagon interaction with the glucagon receptor. Analogues V and VI were found to be partial agonists with maximum stimulation of 7% and 9% and p A_2 values of 6.20 and 6.10, respectively.

Compounds VII and VIII that include the des-His1, Glu⁹, and Tyr⁵ modifications in addition to a D-Tyr¹⁰ or a ${\hbox{\,{\sc D-Phe}}}^{10}$ substitution turned out to be somewhat weak antagonists with binding potency IC₅₀ values of 74.0 and 34.5 nM, respectively, but both were unable to activate the adenylate cyclase system even at concentrations as high as 10^{-5} M. Introducing a D-Phe and its (2R,3R)and (2R,3S)- β -methylphenylalanine derivatives at position 10 decreases considerably the binding affinities. The glucagon analogue VIII turned out to be a weak antagonist unable to activate the adenylate cyclase system even at concentrations as high as 10⁻⁵ M with a p A_2 value of 6.20. The other two β -MePhe isomers, 2R, 3R and 2R, 3S, compounds **X** and **XI** (Table 2), were found to have binding potencies of 120.0 and 180.0 nM, respectively. Analogue X was found to be an antagonist with a p A_2 value of 6.20, whereas analogue **XI** was found to be a partial agonist/antagonist with maximum stimulation of 5.4% and a p A_2 value of 6.15.

Discussion

In this report we have focused on using the des-His¹ and Glu⁹ modifications²⁶ in conjunction with modifications in position 5 and 10 that are known to effect the binding and transduction processes of glucagon with its receptor.^{10,29,30} Threonine in position 5 was replaced by tyrosine which earlier had been shown to aid in producing glucagon antagonists³² with increased potency.³³ In addition, the aromatic amino acid residue Tyr¹⁰ was substituted with Phe¹⁰ and Ala¹⁰ as well as with their

configurational isomers D-Phe¹⁰ and D-Ala¹⁰ to examine what effect these changes in structure and overall orientation of the N-terminal region of glucagon would have on glucagon binding and transduction processes with respect to the glucagon receptor. It has been suggested earlier by Gratzer and Beaven³⁴ that Tyr¹⁰ has a concentration-dependent p K_a which led Korn and Ottensmeyer³⁵ to propose a role for this residue in glucagon action. According to their proposal the Tyr¹⁰ hydroxyl group is hydrogen-bonded to serine in position 8. The losses in binding in going from analogues II to III and then to IV (Table 2) may be due in part to the removal of the phenolic hydroxyl group at that position, but it appears that steric effects of the aromatic ring in position 10 play a major role (Table 2).

According to NMR studies of the glucagon conformation in the presence of lipid micelles, ³⁶ the 10–14 region can contain an α -helical turn in this medium. The nature of this turn, where a hydrophobic patch is created by the Phe⁶, Tyr¹⁰, and Tyr¹³ side chains, is consistent with the X-ray crystallographic structure of glucagon.³⁷ Such a continuous hydrophobic cluster may play an important role in the hormone-receptor interaction. In the case of analogues VII and VIII, though the aromatic ring has been retained, D-substitution should alter the alignment of the Phe⁶, Tyr¹⁰, and Tyr¹³ side chains, which may result in the loss of the continuous hydrophobic region of the hormone. The reduced binding potencies for analogues VII and VIII could be explained by this change in hormone-receptor interaction. For analogues IV and IX, where the aromatic ring has been removed, the binding potency is further decreased due to the reduced hydrophobic interaction with the receptor. It is important to note that most of the analogues which contain L-Ala or a D-amino acid in position 10 have no apparent agonist activity, whereas analogues with L-Phe¹⁰ or L-Tyr¹⁰ do have weak partial agonist activity. These results suggest that the presence of an aromatic side chain in position 10 and its proper orientation are important for biological transduction. Analogues II, III, V, and VI were found to be partial agonists, and their maximal stimulation ranged between 5% and 10% (Table 2). This clearly indicates that agonists favor the L-configuration in position 10, while D-substitutions generally produce pure antagonism. The only D-Phe¹⁰-containing analogue which shows weak stimulation of adenylate cyclase activity is **XI**, which actually has a D- β -MePhe¹⁰ substituent.

The β -methyl amino acids (Figure 1) were initially designed as tools to favor one of three possible staggered rotamers of the side-chain torsion angles χ^1 (Figure 2). It was presumed that the β -methyl substituent, due to its steric repulsion with the adjacent amide and carbonyl groups, will favor the rotamer which places the much less bulky β -proton between the NH and CO groups (Figure 3). Recent NMR and theoretical studies³⁸⁻⁴⁰ revealed a more complicated rotamer equilibrium for the β -methyl-substituted phenylalanine side chains. The side-chain rotamer populations obtained for the β -MePhe stereoisomers in $[\beta$ -MePhe³]CCK-8³⁸ and [β-Me-p-NO₂Phe⁴]DPDPE³⁹ analogues showed that the β -methyl substituent does not always favor one particular χ^1 rotamer but rather always eliminates the χ^1 rotamer which places both bulky β -substituents in a

Figure 1. Structures of the four isomers of β -methylphenylalanine which were incorporated at position 10 of glucagon.

(2R, 3S)-β-MePhe

(2S, 3R)-β-MePhe

Figure 2. *Gauche* (–), *trans*, and *gauche* (+) conformations of α -amino acids.

gauche orientation to the carbonyl group, i.e., a trans rotamer ($\chi^1 \approx 180^\circ$) of both erythro-isomers 2S,3S and 2R,3R, a gauche (+) rotamer ($\chi^1 \approx 60^\circ$) of the (2S,3R)-isomer, and a gauche (–) rotamer ($\chi^1 \approx -60^\circ$) of the 2R,3S-isomer, while allowing the other two conformers in each case.

Probable "bioactive" conformations of the phenylalanine side chains were recently suggested for two classes of β -MePhe-substituted cyclic opioids^{39,40} based on comparative analysis of binding affinities and rotamer preferences of β -MePhe stereoisomers. Although an extensive conformational analysis has not been performed for the glucagon analogues in this study, some preliminary conclusions may be drawn from their relative binding potencies, assuming that the rotamer preferences of β -MePhe¹⁰ stereoisomers are qualitatively similar to that found for β -MePhe side chains in the CCK-8³⁸ and DPDPE³⁹ analogues. The rationale for this assumption is that, despite different positions and, perhaps, different backbone conformations of the β -Me-Phe residues, very similar rotamer populations were found for the same stereoisomers in $[\beta$ -MePhe³]CCK- 8^{38} and [β -Me-p-NO₂Phe⁴]DPDPE analogues.³⁹ In both series of peptides, gauche (-) and gauche (+) rotamers were most populated for (2S,3S)- β -MePhe, and gauche (-) and *trans* rotamers were most populated for (2S,3R)- β -MePhe. For the glucagon series we shall relate binding potencies of β -MePhe¹⁰-containing analogues to the L-Phe¹⁰ substitution, analogue III, which possesses almost the same binding affinity as the parent antagonist [des-His¹,Tyr⁵,Glu⁹]glucagon amide containing tyrosine in position 10. The (2S,3R)- β -MePhe¹⁰ stereoisomer V was only 2 times less potent than the reference analogue III, while the 2S,3S-stereoisomer VI was 37 times less potent. This indicates that the additional β -methyl group does not significantly affect the receptor

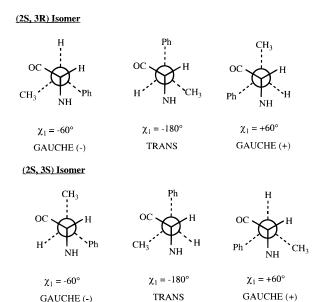


Figure 3. Conformational analysis of the β -methylphenylalanine 2S, 3R- and 2S, 3S-diastereomers.

binding of the 2S,3R-stereoisomer but considerably affects the binding of the 2S,3S-stereoisomer. The gauche (+) rotamer of β -MePhe¹⁰, according to previous results,^{38,39} is expected to be more populated for the less potent 2*S*,3*S*-stereoisomer **VI** than for the more potent 2S,3R-stereoisomer **V**. Therefore, this rotamer population seems to be irrelevant in the receptor binding of the glucagon antagonists. In contrast, the trans rotamer, which was shown to be more populated for the 2*S*,3*R*-stereoisomer than for the 2*S*,3*S*-stereoisomer, seems to be the most probable "binding" rotamer of phenylalanine side chains in the L-Phe¹⁰-containing analogues III, V, and VI. At present we cannot exclude the alternative possibility that the phenylalanine side chains bind to the receptor in a gauche (-) orientation, which is expected to be quite populated for all three L-Phe¹⁰-containing analogues (III, V, and VI), and that the low affinity of the 2*S*,3*S*-stereoisomer is caused by direct interference of its β -methyl group with the receptor binding site.

The relatively low binding potencies of the D-Phe¹⁰-containing analogues **VIII**, **X**, and **XI** may be attributed in the first place to a local change in backbone conformation imposed by the D-residue substitution. Note that the D-Phe¹⁰ analogue **VIII**, which has no modifications at the β -carbon, is 9 times less potent than the reference L-Phe¹⁰-containing analogue **III** (Table 2). The possible changes in hydrophobic stacking of Phe⁶, Tyr¹⁰, and Tyr¹³ upon a D-amino acid substitution have been discussed above. These changes combined with a reduced side-chain mobility may cause a bad fit of analogues **X** and **XI** to the glucagon receptor, which resulted in their poor binding affinities.

In summary, 10 new glucagon analogues have been designed and synthesized in order to explore structural and topographical requirements of the glucagon receptor to the amino acid in position 10 of the glucagon sequence. The substitution of L-Phe for L-Tyr shows that the hydroxyl group in position 10 has only a weak influence on the receptor binding. However, introducing a D-Phe and its β -methyl derivatives into position 10 considerably decreases the binding affinity, presumably because of an influence on conformation preferred by

Table 3. Physical Data for Glucagon Analogues

	M	W		TLC , c R_{f}		
peptide	calcd	found ^a	HPLC K^{b}	A	В	С
II	3420.79	3420.60	4.15	0.65	0.63	0.62
III	3404.80	3404.90	4.18	0.66	0.61	0.64
IV	3328.70	3328.30	4.01	0.66	0.73	0.69
V	3418.80	3419.20	4.37	0.49	0.55	0.47
VI	3418.80	3418.60	3.89	0.53	0.53	0.50
VII	3420.79	3420.50	4.56	0.63	0.64	0.59
VIII	3404.80	3404.90	4.05	0.60	0.62	0.64
IX	3328.70	3329.10	4.61	0.58	0.61	0.63
X	3418.80	3419.20	4.21	0.51	0.49	0.54
XI	3418.80	3418.10	4.07	0.58	0.59	0.58

^a MW found by electrospray mass spectroscopy. ^b K values were determined under the following conditions: gradient, 10−90% acetonitrile in 0.1% TFA over 30 min; flow rate, 1.5 mL/min. ^c TLC: silica gel 60 F-245, 0.25 mm layer thickness; solvent systems: A, 1-butanol/acetic acid/pyridine/water (5:4:1:4); B, 1-butanol/acetic acid/pyridine/water (4:1:1:3); C, ethyl acetate/pyridine/acetic acid/water (12:4:4.2:2.2).

the glucagon receptor. The 15-fold difference between binding potencies of the (2S,3R)- and (2S,3S)- β -MePhe¹⁰ analogues reveals strong topographical requirements of the glucagon receptor to the orientation of the phenyl ring in position 10. Presently we cannot discriminate between the *trans* and *gauche* (–) rotamers as the possible phenylalanine orientations at the receptor. Further structure—activity and conformational studies are necessary to understand the glucagon receptor requirements.

Experimental Section

Materials. N^{α} -Fmoc- and N^{α} -Boc-protected amino acids, pMBHA resin (0.51 mmol of NH₂/g substitution), and 4-[(2',4'dimethoxyphenyl)-Fmoc-(aminomethyl)]phenoxy resin were purchased from Bachem (Torrence, CA). Other chemicals and solvents were purchased from the following sources: trifluoroacetic acid (TFA) (Halocarbon Products, NJ); anisole, 1,2ethanedithiol, 1-hydroxybenzotiazole (HOBt), diisopropylethylamine (DIEA), diisopropylcarbodiimide (DIC), and 1-methyl-2-pyrrolidinone (NMP) (Aldrich, Milwaukee, WI); dichloromethane (DCM) (Mallinckrodt Specialty Chemicals, Paris, KY); HPLC-quality acetonitrile (Burdick & Jackson, Muskegon, MI); (benzotriazolyloxy)tris(dimethylamino)phosphonium hexafluorophosphate (BOP reagent) (Peptide International, Louiville, KY); 2-(1*H*-benzotrizol-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate (HBTU) (Richelieu Biotechnologies Inc., Saint Hyacinthe, Canada); bovine serum albumin, cAMP, GTP, ATP, chromatographic alumina (type WN3, neutral), and all enzymes (Sigma Chemicals, St. Louis, MO); [125I]glucagon, [3H]cAMP, and α -32P (New England Nuclear, Boston, MA); N,N-dimethylformamide (DMF) (Fischer Scientific, Pittsburgh, PA); and Dowex AG 50-W4 cation exchange resin (BioRad, San Diego, CA). TLC was performed using Merck silica gel 60 F-254 plates (0.25 mm layer thickness), and the following solvents were used (Table 3): (A) 1-butanol/acetic acid/water/ pyridine (5:4:4:1); (B) 1-butanol/acetic acid/water/pyridine (4: 1:3:1); (C) ethyl acetate/pyridine/acetic acid/water (12:4:4.2: 2.2). The peptides were detected on the TLC plates using iodine vapor. The purification was achieved on a Perkin-Elmer Model 410-BIO instrument by a preparative high-performance liquid chromatography (HPLC) C₁₈-bonded silica gel column (VYDAC 218 TBP-16, 16×250 mm). The peptides were eluted with a linear gradient of acetonitrile in 0.1% aqueous TFA (10-90%) over 30 min at a flow rate of 5.0 mL/min. The separations were monitored at 280 nm and integrated with a Perkin-Elmer LC-235 diode array detector. The amino acid analyses were done at the University of Arizona Biotechnology Core Facility. The system used was an Applied Biosystems Model 420A amino acid analyzer with automatic hydrolysis (vapor phase at 160 °C for 1 h 40 min using 6 N HCl) and precolumn phenylthiocabamyl-amino acid (PTC-AA) analysis.

The purity of the peptides was checked by analytical RP-HPLC using a VYDAC 218 TBP-16 column (4.6×250 mm) and by TLC in three different solvent systems (Table 3). The structures of the pure peptides were confirmed by electrospray mass spectroscopy (Table 3) and amino acid analysis.

Synthesis of the N^{α} -tert-Butyloxycarbonyl Derivatives of (2*S*,3*S*)- and (2*S*,3*R*)- β -Methylphenylalanine. The syntheses of N^{α} -(tert-butyloxycarbonyl)- β -methylphenylalanine isomers were carried out by slight changes of previously published experiments.⁴¹ Briefly, a solution of 2.0 g (11.2 mmol) of (2S,3S)- β -MePhe or 2.0 g (11.2 mmol) of (2S,3R)- β -MePhe in 100 mL of dioxane-water (2:1) was cooled in an ice bath, and 2.70 g (12.4 mmol) of di-tert-butyl dicarbonate was added slowly. The pH was adjusted to 10.5 with 5 N sodium hydroxide, and the solution was continuously stirred and adjusted to 10.5. The mixture was stirred at room temperature overnight. The solvents were removed under vacuum, the solid was dissolved in water and cooled to 0 $^{\circ}\text{C},$ and 70 mL of ethyl acetate was added. The aqueous solution was cooled to 0 °C, acidified with 10% citric acid, and extracted with 4×50 mL portions of ethyl acetate. The collected organic phases were washed twice with water and brine, dried over sodium sulfate, and evaporated to dryness. The white crystal product finally gave 2.25 g of N^{t_1} -Boc-(2S,3S)- β -MePhe (71%) and 2.18 g of N^{α} -Boc-(2S,3R)- β -MePhe (69%) whose properties are the same as those previously reported.

Peptide Synthesis. (A) Using an N^{α} -Fmoc strategy, eight of the glucagon analogues were synthesized by solid-phase methodology using a 4-[(2',4'-dimethoxyphenyl)-Fmoc-(aminomethyl)]phenoxy resin. 42 The synthesis of the glucagon analogues [des-His1, Tyr5, Glu9] glucagon-NH2 (II), [des-His1,-Tyr⁵,Glu⁹,Phe¹⁰]glucagon-NH₂ (III), [des-His¹,Tyr⁵,Glu⁹,Ala¹⁰]glucagon-NH₂ (IV), [des-His¹,Tyr⁵,Glu⁹,D-Tyr¹⁰]glucagon-NH₂ **(VII)**, [des-His¹,Tyr⁵,Glu⁹,D-Phe¹⁰]glucagon-NH₂ **(VIII)**, [des-His¹,Tyr⁵,Glu⁹,D-Ala¹⁰]glucagon-NH₂ **(IX)**, [des-His¹,Tyr⁵,Glu⁹,-(2R, 3R)- β -MePhe¹⁰]glucagon-NH₂ (**X**), and [des-His¹,Tyr⁵,-Glu⁹,(2R,3S)- β -MePhe¹⁰]glucagon-NH₂ (**XI**) were carried out on an Applied Biosystems (ABI) 431A automated synthesizer on a 0.25 mmol scale. The following side-chain-protecting groups were used Arg(2,2,5,7,8-pentamethylchroman-6-sulfonyl,Pmc), Asn(trityl), Asp(t-Bu), Gln(trityl), Glu(t-Bu), Lys-(Boc), Ser(t-Bu), Thr(t-Bu), and Tyr(t-Bu). N^{α} -Fmoc-protected amino acids (4 equiv) were added sequentially, using HBTU with HOBt as coupling reagents in NMP. The analogues were cleaved from the resin using standard techniques and a cleavage mixture of 90% trifluoroacetic acid, 5% anisole, 2.5% methyl sulfide, and 2.5% 1,2-ethanedithiol.

(B) Using an N^{L} -Boc strategy, the other two glucagon analogues, [des-His¹,Tyr⁵,Glu⁹,(2S,3S)- β -MePhe¹⁰]glucagon-NH₂ (**V**) and [des-His¹,Tyr⁵,Glu⁹,(2S,3R) β -MePhe¹⁰]glucagon-NH₂ (VI), were synthesized using a Vega Coupler 250 or a Vega Coupler 1000 synthesizer. These two compounds were prepared using p-methylbenzhydrylamine resin substituted at the level of 0.51 mmol/g of resin. The following amino acid side-chain-protecting groups were used ${\rm Arg}(N^G{\text{-}Tos})$, Asp-(O-Bzl), $Glu(\gamma$ -O-Bzl), Lys(N^{ϵ} -2,4-diCl-Cbz), Ser(O-Bzl), Thr-(O-Bzl), Trp(Nin-For), and Tyr(2,6-diCl-Bzl). A sample of pMBHA resin was neutralized with 10% DIEA in DCM (2 imes50 mL) followed by a DCM wash (4 \times 50 mL). The amino acid N^{α} -Boc-Thr(Bzl) was coupled to the resin using DIC and HOBt as coupling reagents for 2 h. N^{α} -Boc amino acids were added to the growing peptide chain as their preformed symmetrical anhydrides in 3-fold excess. The synthetic protocol consisted of the following steps: (a) 50% TFA in DCM with 1% methionine (1 \times 2 min); (b) same as step a (1 \times 2 min); (c) DCM (3 \times 2 min); (d) 10% DIEA in DMF (3 \times 2 min); (e) DCM $(3 \times 2 \text{ min})$; (f) coupling by preformed symmetric anhydride in DCM for 30 min or by DIC/HOBt for 2 h; (g) 10% DIEA in DCM (1 \times 2 min); (h) DCM (1 \times 2 min); (i) coupling by preformed symmetric anhydride in DMF for 30 min or by DIC/ HOBt for 2 h; (k) DCM (3 \times 2 min); (l) ethanol (3 \times 2 min). The reactions were monitored by the Kaiser test.³¹ After the last synthetic cycle, the N-terminal Boc group was removed (steps a-e) and the resin was dried in vacuo. The low/high-HF technique was used to cleave the peptide from the resin to avoid any side reactions.43

Isolation of Liver Plasma Membrane. The liver plasma membranes were obtained from Sprague–Dawley rats weighing 200-250 g as described by Neville⁴⁴ and modified by Pohl.⁴⁵ The protein concentration was determined by the method of Markwell *et al.*⁴⁶

Receptor Binding Assay. The binding assay was performed according to Wright and Rodbell⁴⁷ in which competition for glucagon receptors between [125I]glucagon and the glucagon analogue was measured. Briefly, an incubation medium had a volume of 500 μ L consisting of liver plasma membrane containing 50 μ g of protein, 15×10^4 cpm of [125I]glucagon, and unlabeled glucagon or glucagon analogues at a desired concentration (range from 0 to 10 μ M), all in 25 mM Tris-HCl with 0.5% BSA (pH 7.4 at 25 °C). The mixture was incubated for 10 min at 30 °C followed by immediate cooling in an ice bath and filtered through a 0.45 μm cellulose acetate filter previously soaked for 12 h in a Tris-BSA buffer; 4 mL of ice cold Tris-BSA buffer was used for washing, and the amount of radioactivity remaining on the filter was quantitated using a LKB1275 mini-gamma counter. Nonspecific binding, measured in the presence of excess unlabeled peptide (1024 nM), was 15-20% of the total binding and substracted in each case to give the specific binding. Results are expressed as the percent inhibition of [125I]glucagon specific binding. Assays were performed in triplicate and repeated twice.

Adenylate Cyclase Assay. The adenylate cyclase activity was measured by the conversion of $[\alpha^{-32}P]ATP$ to 3′,5′-cyclic AMP as described by Lin $et~al.^{21}~$ Labeled cAMP was determined by the method of Solomon $et~al.^{48}$ using Dowex 50 and alumina columns. Briefly, 0.1 mL of incubation medium consisted of 1 mM $[\alpha^{-32}P]ATP~$ ($\sim 50~$ cpm/pmol), 5 mM MgCl $_2$, 10 μ M GTP, 1 mM EDTA, 1 mM cAMP, containing 10⁴ cpm of $[^3H]cAMP,$ 25 mM Tris-HCl (pH 7.4 at 25 °C), 1% BSA, 35 μ g of membrane protein, and an ATP regeneration system that had 20 mM phosphocreatine and 0.72 mg/mL (100 units/mL) creatine phosphokinase. Results are expressed as the percent of stimulation of cAMP production over basal. Assays were done twice in triplicate.

For the pA_2 values, a dose–response plot was obtained by determining the response when the concentration of glucagon was varied while the concentration of the antagonist was kept the same. The plots were generated by using different concentrations ranging between 1 and $10~\mu M$ antagonist. The dose–response curves exhibit shifts in the EC_{50} values which were used in calculating the pA_2 values as described by Schild.⁴⁹

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